**Note:** *For the benefit of the students, specially the aspiring ones, the question of JEE(Main), 2016 are also given in this booklet. Keeping the interest of students studying in class XI, the questions based on topics from class XI have been marked with '*', which can be attempted as a test.*

# FIITJEE

# Solutions to JEE(Main)-2016

Test Booklet Code

**F**

**PAPER – 1 CHEMISTRY, MATHEMATICS & PHYSICS**

***Important Instructions:***

1. Immediately fill in the particular on this page of the Test Booklet with ***only Blue / Black Ball Point Pen*** provided by the Board.
2. The Answer Sheet is kept inside this Test Booklet. When you are directed to open the Test Booklet, take out the Answer Sheet and fill in the particulars carefully.
3. The test is of **3 hours** duration.
4. The Test Booklet consists of **90** questions. The maximum marks are **360**.
5. There are ***three*** parts in the question paper A, B, C consisting of **Chemistry, Mathematics** and **Physics** having 30 questions in each part of equal weightage. Each question is allotted **4 (four)** marks for each correct response.
6. *Candidates will be awarded marks as started above in instruction No. 5 for correct response of each question. ¼ (one fourth) marks will be deducted for indicating incorrect response of each question. No deduction from that total score will be made if no response is indicated for an item in the answer sheet.*
7. There is only one correct response for each question. Filling up more than one response in each question will be treated as wrong response and marks for wrong response will be deducted accordingly as per instruction 6 above.
8. For writing particulars / marking responses on ***Side–1*** and ***Side–2*** of the Answer Sheet use ***only Blue/Black Ball Point Pen*** provided by the Board.
9. No candidate is allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, any electronic device, etc. except the Admit Card inside the examination room/hall.
10. Rough work is to be done on the space provided for this purpose in the Test Booklet only. This space is given at the bottom of each page.
11. On completion of the test, the candidate must hand over the Answer Sheet to the Invigilator on duty in the Room / Hall. ***However, the candidates are allowed to take away this Test Booklet with them***.
12. The CODE for this Booklet is **F**. Make sure that the CODE printed on **Side–2** of the Answer Sheet and also tally the serial number of the Test Booklet and Answer Sheet are the same as that on this booklet. In case of discrepancy, the candidate should immediately report the matter to the Invigilator for replacement of both the Test Booklet and the Answer Sheet.
13. **Do not fold or make any stray mark on the Answer Sheet.**

Name of the Candidate (in Capital letters): ____________________

Roll Number : in figures [ | | | | | | | | ]

: in words ____________________

Examination Centre Number : [ | | | | | | ]

Name of Examination Centre (in Capital letters): ____________________

Candidate's Signature : ____________________

1. Invigilator's Signature : ______________

2. Invigilator's Signature : ______________

# PART A – CHEMISTRY

1*. A stream of electrons from a heated filament was passed between two charged plates kept at a potential difference *V* esu. If *e* and *m* are charge and mass of an electron, respectively, then the value of $h/\lambda$ (where $\lambda$ is wavelength associated with electron wave) is given by :

(1) 2meV (2) $\sqrt{meV}$
(3) $\sqrt{2meV}$ (4) meV

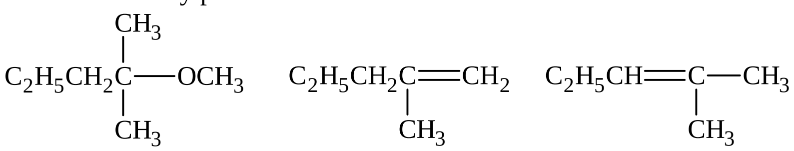
***Sol.*** **(3)**

$$\lambda = \frac{h}{mv}$$

$$= \frac{h}{\sqrt{2m(KE)}}$$

$$\therefore \quad \frac{h}{\lambda} = \sqrt{2meV}$$

2. 2-chloro-2-methylpentane on reaction with sodium methoxide in methanol yields :



(a) (b) (c)

(1) (a) and (c) (2) (c) only
(3) (a) and (b) (4) All of these

***Sol.*** **(4)**
via $S_N1$ and $E_2$ mechanisms

3. Which of the following compounds is metallic and ferromagnetic ?
(1) $CrO_2$ (2) $VO_2$
(3) $MnO_2$ (4) $TiO_2$

***Sol.*** **(1)**
$CrO_2$ is metallic and ferromagnetic.

4. Which of the following statements about low density polythene is FALSE ?
(1) It is a poor conductor of electricity.
(2) Its synthesis requires dioxygen or a peroxide initiator as a catalyst.
(3) It is used in the manufacture of buckets, dust-bins etc.
(4) Its synthesis requires high pressure.

***Sol.*** **(3)**
HDPE is used in making buckets and dustbins

5. For a linear plot of log (x/m) versus log p in a Freundlich adsorption isotherm, which of the following statements is correct ? (k and n are constants)
(1) 1/n appears as the intercept. (2) Only 1/n appears as the slope
(3) log (1/n) appears as the intercept. (4) Both k and 1/n appear in the slope term.

***Sol.*** **(2)**

$$\frac{x}{m} = kp^{\frac{1}{n}}$$

$$\log\frac{x}{m} = \log k + \frac{1}{n}\log p$$

6*. The heats of combustion of carbon and carbon monoxide are –393.5 and –283.5 kJ mol$^{-1}$, respectively. The heat of formation (in kJ) of carbon monoxide per mole is :

(1) 676.5 (2) –676.5
(3) –110.5 (4) 110.5

***Sol.*** **(3)**

$$C + O_2 \longrightarrow CO_2 \qquad \ldots(1)$$

$$CO + \frac{1}{2}O_2 \longrightarrow CO_2 \qquad \ldots(2)$$

$$\Delta H_1 \Rightarrow -393.5 \text{ kJ / mole}$$

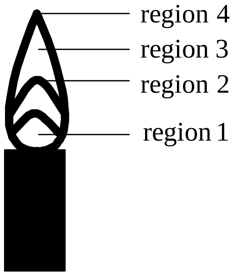
$$\Delta H_2 \Rightarrow -283.5 \text{ kJ / mole}$$

$$C + \frac{1}{2}O_2 \longrightarrow CO$$

$$\Delta_f H_{(CO,\,g)} = \Delta H_1 - \Delta H_2$$

$\Rightarrow$ –393.5 –(–283.5)
= –393.5 + 283.5
$\Rightarrow$ –110 kJ/mole

7. The hottest region of Bunsen flame shown in the figure below is :




(1) region 2 (2) region 3
(3) region 4 (4) region 1

***Sol.*** **(1)**
region 2

8. Which of the following is an anionic detergent ?

(1) Sodium lauryl sulphate (2) Cetyltrimethyl ammonium bromide
(3) Glyceryl oleate (4) Sodium stearate

***Sol.*** **(1)**
Sodium $^{+}$(lauryl sulphate)$^{-}$ is an anionic detergent.

9. 18 g glucose ($C_6H_{12}O_6$) is added to 178.2 g water. The vapour pressure of water (in torr) for this aqueous solution is :

(1) 76.0 (2) 752.4
(3) 759.0 (4) 7.6

***Sol.*** **(2)**

$n_{C_6H_{12}O_6} = 0.1$

$n_{H_2O} = \frac{178.2}{18} = 9.9$

$\frac{p^0 - p}{p^0} \Rightarrow \frac{0.1}{10}$

$p^0 - p \Rightarrow \frac{0.1}{10} \times 760 = 7.6$

$p = p^0 - 7.6$

= 760–7.6

= 752.4 torr

10*. The distillation technique most suited for separating glycerol from spent-lye in the soap industry is :

(1) Fractional distillation
(2) Steam distillation
(3) Distillation under reduced pressure
(4) Simple distillation

***Sol.*** **(3)**

Glycerol decomposes before its boiling point.

11*. The species in which the N atom is in a state of *sp* hybridization is :

(1) $NO_2^-$
(2) $NO_3^-$
(3) $NO_2$
(4) $NO_2^+$

***Sol.*** **(4)**

$O = \overset{\oplus}{N} = O$

12. Decomposition of $H_2O_2$ follows a first order reaction. In fifty minutes the concentration of $H_2O_2$ decreases from 0.5 to 0.125 M in one such decomposition. When the concentration of $H_2O_2$ reaches 0.05 M, the rate of formation of $O_2$ will be :

(1) $6.93 \times 10^{-4}$ mol min$^{-1}$
(2) 2.66 L min$^{-1}$ at STP
(3) $1.34 \times 10^{-2}$ mol min$^{-1}$
(4) $6.93 \times 10^{-2}$ mol min$^{-1}$

***Sol.*** **(1)**

$H_2O_2 \longrightarrow H_2O + \frac{1}{2}O_2$

$2t_{1/2} = 50$ mins

$t_{1/2} = 25$ mins

$\therefore\ k = \frac{0.693}{25}$

$-\frac{d[H_2O_2]}{dt} = \frac{d[O_2]}{dt} \times 2$

$\therefore\ \frac{d[O_2]}{dt} = \frac{1}{2} \times \frac{0.693}{25} \times 0.05$

$= 6.93 \times 10^{-4}$

13. The pair having the same magnetic moment is :

[At. No.: Cr = 24, Mn = 25, Fe = 26, Co = 27]

(1) $[Cr(H_2O)_6]^{2+}$ and $[Fe(H_2O)_6]^{2+}$
(2) $[Mn(H_2O)_6]^{2+}$ and $[Cr(H_2O)_6]^{2+}$
(3) $[CoCl_4]^{2-}$ and $[Fe(H_2O)_6]^{2+}$
(4) $[Cr(H_2O)_6]^{2+}$ and $[CoCl_4]^{2-}$

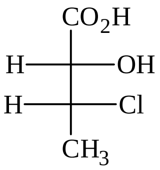
***Sol.*** **(1)**

$Cr^{2+} \longrightarrow d^4 \longrightarrow$ No of unpaired electron 4

$Fe^{2+} \longrightarrow d^6 \longrightarrow$ No of unpaired electron 4

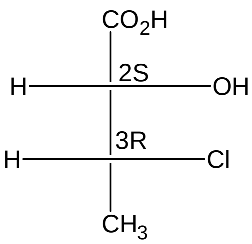
Both configurations have 4 unpaired electrons due to weak field ligands.

14*. The absolute configuration of




(1) (2S, 3R)
(2) (2S, 3S)
(3) (2R, 3R)
(4) (2R, 3S)

***Sol.*** **(1)**




15*. The equilibrium constant at 298 K for a reaction $A + B \rightleftharpoons C + D$ is 100. If the initial concentration of all the four species were 1M each, then equilibrium concentration of D (in mol $L^{-1}$) will be:

(1) 0.818
(2) 1.818
(3) 1.182
(4) 0.182

***Sol.*** **(2)**

| $A$ | $+$ $B \rightleftharpoons$ | $C$ | $+$ $D$ |
|---|---|---|---|
| 1 | 1 | 1 | 1 |
| 1–x | 1–x | 1+x | 1+x |

Given $K_{eq} = 100$

$$\frac{(1+x)^2}{(1-x)^2} = 100$$

$$\frac{1+x}{1-x} = 10$$

$$1 + x = 10 - 10x$$

$$11x = 9$$

$$x = \frac{9}{11}$$

$$[D] = 1 + \frac{9}{11}$$

$$= 1 + 0.818$$

= 1.818

16. Which one of the following ores is best concentrated by froth floatation method?
(1) Siderite (2) Galena
(3) Malachite (4) Magnetite

***Sol.*** **(2)**
Galena (PbS) is a sulphide ore.

17*. At 300 K and 1 atm, 15 mL of a gaseous hydrocarbon requires 375 mL air containing 20% $O_2$ by volume for complete combustion. After combustion the gases occupy 330 mL. Assuming that the water formed is in liquid form and the volumes were measured at the same temperature and pressure, the formula of the hydrocarbon is:
(1) $C_3H_8$ (2) $C_4H_8$
(3) $C_4H_{10}$ (4) $C_3H_6$

***Sol.*** **None**
Volume of $O_2$ = $375\times\frac{20}{100}$ = 75 ml
Volume of gaseous hydrocarbon = 15 ml

$$C_xH_y + \left(x+\frac{y}{4}\right)O_2 \longrightarrow xCO_2 + \frac{y}{2}H_2O(\ell)$$

$$15\left(x+\frac{y}{4}\right) = 75 \qquad \ldots(1)$$

$$x+\frac{y}{4} = 5$$

Excess air = 375 – 75 = 300
300 + volume of $CO_2$ = 330
Volume of $CO_2$ = 30
15 x = 30
x = 2
(Correct option is not given)

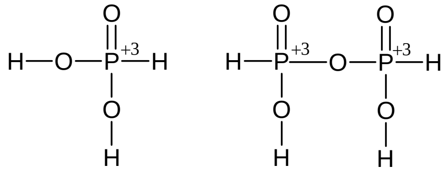
$$2+\frac{y}{4} = 5$$

$$\frac{y}{4} = 3$$

y = 12
$C_2H_{12}$ (not possible)
Hence No answer is correct.
**However, considering the equation no. 1 alone, if we put x and y values, 3 and 8 respectively, then equation no. 1 is satisfied and answer will be $C_3H_8$.**

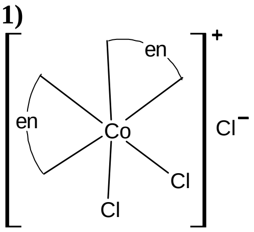
18. The pair in which phosphorous atoms have a formal oxidation state of +3 is:
(1) Pyrophosphorous and hypophosphoric acids
(2) Orthophosphorus and hypophosphoric acids
(3) Pyrophosphorous and pyrophosphoric acids
(4) Orthophosphorous and pyrophosphorous acids

***Sol.*** **(4)**
$H_4P_2O_5 \Rightarrow$ pyrophosphorous (+ 3 state)
$H_3PO_3 \Rightarrow$ orthophosphorous (+ 3 state)



19. Which one of the following complexes shows optical isomerism?
(1) *cis*$[Co(en)_2Cl_2]Cl$
(2) *trans*$[Co(en)_2Cl_2]Cl$
(3) $[Co(NH_3)_4Cl_2]Cl$
(4) $[Co(NH_3)_3Cl_3]$
(en = ethylenediamine)

***Sol.*** **(1)**




$cis\left[Co(en)_2Cl_2\right]Cl$
(optically active)

20. The reaction of zinc with dilute and concentrated nitric acid, respectively, produces:
(1) $NO_2$ and NO
(2) NO and $N_2O$
(3) $NO_2$ and $N_2O$
(4) $N_2O$ and $NO_2$

***Sol.*** **(4)**

$$Zn + \underset{conc.}{4HNO_3} \longrightarrow Zn\left(NO_3\right)_2 + 2H_2O + +2NO_2 \uparrow$$

$$4Zn + \underset{dil.}{10HNO_3} \longrightarrow 4Zn\left(NO_3\right)_2 + N_2O \uparrow + +5H_2O$$

21*. Which one of the following statements about water is **FALSE**?
(1) Water can act both as an acid and as a base.
(2) There is extensive intramolecular hydrogen bonding in the condensed phase.
(3) Ice formed by heavy water sinks in normal water
(4) Water is oxidized to oxygen during photosynthesis

***Sol.*** **(2)**
Ice shows intermolecular H – bonding.

22*. The concentration of fluoride, lead, nitrate and iron in a water sample from an underground lake was found to be 1000 ppb, 40 ppb, 100 ppm and 0.2 ppm, respectively. This water is unsuitable for drinking due to high concentration of :
(1) Lead
(2) Nitrate
(3) Iron
(4) Fluoride

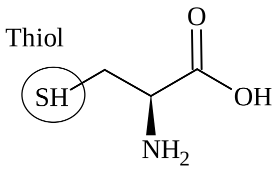
***Sol.*** **(2)**
Maximum limit of nitrate in potable water is 50 ppm.

23*. The main oxides formed on combustion of Li, Na and K in excess of air are, respectively:
(1) $LiO_2$, $Na_2O_2$ and $K_2O$
(2) $Li_2O_2$, $Na_2O_2$ and $KO_2$
(3) $Li_2O$, $Na_2O_2$ and $KO_2$
(4) $Li_2O$, $Na_2O$ and $KO_2$

***Sol.*** **(3)**

24. Thiol group is present in :
(1) Cystine
(2) Cysteine
(3) Methionine
(4) Cytosine

***Sol.*** **(2)**




25. Galvanization is applying a coating of:
(1) Cr (2) Cu
(3) Zn (4) Pb

***Sol.*** **(3)**

26*. Which of the following atoms has the highest first ionization energy?
(1) Na (2) K
(3) Sc (4) Rb

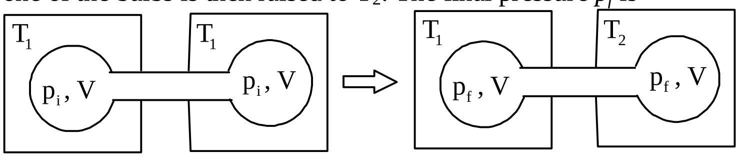
***Sol.*** **(3)**

27. In the Hofmann bromamide degradation reaction, the number of moles of NaOH and $Br_2$ used per mole of amine produced are:
(1) Four moles of NaOH and two moles of $Br_2$.
(2) Two moles of NaOH and two moles of $Br_2$
(3) Four moles of NaOH and one mole of $Br_2$
(4) One mole of NaOH and one mole of $Br_2$

***Sol.*** **(3)**

$$R-\overset{\overset{O}{\parallel}}{C}-NH_2 + 4NaOH + Br_2 \longrightarrow RNH_2 + 2NaBr + Na_2CO_3 + 2H_2O$$

28*. Two closed bulbs of equal volume ($V$) containing an ideal gas initially at pressure $p_i$ and temperature $T_1$ are connected through a narrow tube of negligible volume as shown in the figure below. The temperature of one of the bulbs is then raised to $T_2$. The final pressure $p_f$ is




(1) $2p_i\left(\frac{T_1}{T_1+T_2}\right)$ (2) $2p_i\left(\frac{T_2}{T_1+T_2}\right)$
(3) $2p_i\left(\frac{T_1T_2}{T_1+T_2}\right)$ (4) $p_i\left(\frac{T_1T_2}{T_1+T_2}\right)$

***Sol.*** **(2)**
Initial total moles:

$$n = \frac{p_iV}{RT_1} + \frac{p_iV}{RT_1} = \frac{2p_iV}{RT_1} \quad \ldots (1)$$

Final total moles:

$$n = \frac{p_fV}{RT_1} + \frac{p_fV}{RT_2} \quad \ldots (2)$$

Equating the two:

$$\frac{2p_i}{T_1} = \frac{p_f}{T_1} + \frac{p_f}{T_2}$$

$$p_f = \frac{2p_i}{T_1} \times \frac{T_1 T_2}{(T_1 + T_2)}$$

$$p_f = 2p_i \times \frac{T_2}{T_1 \times T_2}$$

29*. The reaction of propene with HOCl ($Cl_2 + H_2O$) proceeds through the intermediate:

(1) $CH_3 - CH^+ - CH_2 - Cl$

(2) $CH_3 - CH(OH) - CH_2^+$

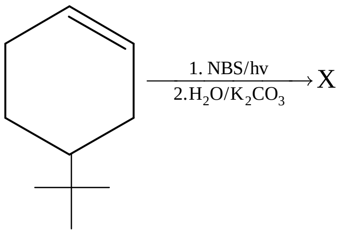
(3) $CH_3 - CHCl - CH_2^+$

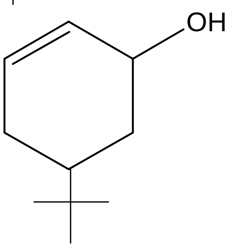
(4) $CH_3 - CH^+ - CH_2 - OH$

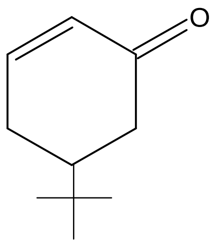
***Sol.*** **(1)**

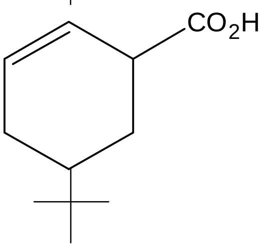
$CH_3 - CH = CH_2 + Cl^{\oplus} \longrightarrow CH_3 - \overset{\oplus}{C}H - CH_2Cl$ (Markovnikov's addition)

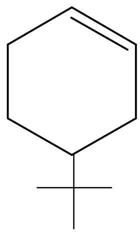
30*. The product of the reaction given below is:




(1) 


(2) 


(3) 


(4) 

***Sol.*** **(1)**
Allylic bromination followed by hydrolysis.

# PART B – MATHEMATICS

*31. Two sides of a rhombus are along the lines, $x - y + 1 = 0$ and $7x - y - 5 = 0$. If its diagonals intersect at $(-1, -2)$, then which one of the following is a vertex of this rhombus?

(1) $(-3, -8)$
(2) $\left(\frac{1}{3}, -\frac{8}{3}\right)$
(3) $\left(-\frac{10}{3}, -\frac{7}{3}\right)$
(4) $(-3, -9)$

***Sol.*** **(2)**
Remaining two sides of rhombus are $x - y - 3 = 0$ and $7x - y + 15 = 0$.
So on solving, we get vertices as $\left(\frac{1}{3}, -\frac{8}{3}\right)$, (1, 2), $\left(-\frac{7}{3}, -\frac{4}{3}\right)$ and $(-3, -6)$.

*32. If the 2nd, 5th and 9th terms of a non–constant A.P. are in G.P., then the common ratio of this G.P. is:

(1) $\frac{4}{3}$
(2) 1
(3) $\frac{7}{4}$
(4) $\frac{8}{5}$

***Sol.*** **(1)**
Let the G.P. be a, ar, $ar^2$ and terms of A.P. are A + d, A +4d, A + 8d
then $\frac{ar^2 - ar}{ar - a} = \frac{(A+8d)-(A+4d)}{(A+4d)-(A+d)} = \frac{4}{3}$
$\Rightarrow r = \frac{4}{3}$.

**Alternate Solution:**
Let AP is a, a + d, a + 2d .....
2nd, 5th and 9th terms a + d, a + 4d, a + 8d are in GP
$\Rightarrow (a + 4d)^2 = (a + d)(a + 8d)$
$\Rightarrow d(8d - a) = 0 \Rightarrow 8d = a$ as $d \neq 0$
Hence common ratio of GP $\frac{a+4d}{a+d} = \frac{8d+4d}{8d+d} = \frac{12d}{9d} = \frac{4}{3}$.

*33. Let P be the point on the parabola, $y^2 = 8x$ which is at a minimum distance from the centre C of the circle, $x^2 + (y + 6)^2 = 1$. Then the equation of the circle, passing through C and having its centre at P is:

(1) $x^2 + y^2 - x + 4y - 12 = 0$
(2) $x^2 + y^2 - \frac{x}{4} + 2y - 24 = 0$
(3) $x^2 + y^2 - 4x + 9y + 18 = 0$
(4) $x^2 + y^2 - 4x + 8y + 12 = 0$

***Sol*.** **(4)**

Equation of normal at P is

$y = -tx + 4t + 2t^3$

It passes through C(0, –6)

$\therefore \quad -6 = 4t + 2t^3$

$\Rightarrow \quad t^3 + 2t + 3 = 0$

$\Rightarrow \quad t = -1$

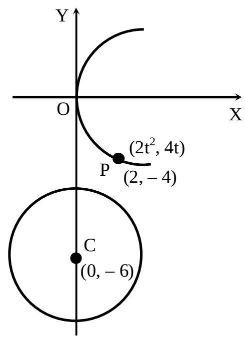
Hence P is (2, – 4)

$r = \sqrt{4+4} = 2\sqrt{2}$

Equation of required circle

$(x-2)^2 + (y+4)^2 = 8$

$\Rightarrow \quad x^2 + y^2 - 4x + 8y + 12 = 0$




34. The system of linear equations

$x + \lambda y - z = 0$

$\lambda x - y - z = 0$

$x + y - \lambda z = 0$

has a non–trivial solution for :

(1) exactly one value of λ.
(2) exactly two values of λ.
(3) exactly three values of λ.
(4) infinitely many values of λ.

***Sol*.** **(3)**

For non-trivial solution

$$\begin{vmatrix} 1 & \lambda & -1 \\ \lambda & -1 & -1 \\ 1 & 1 & -\lambda \end{vmatrix} = 0$$

$1(\lambda + 1) - \lambda(-\lambda^2 + 1) - 1(\lambda + 1) = 0$

$\lambda + 1 + \lambda^3 - \lambda - \lambda - 1 = 0$

$\lambda(\lambda^2 - 1) = 0 \Rightarrow \lambda = 0, \lambda = \pm 1.$

35. If $f(x) + 2f\left(\frac{1}{x}\right) = 3x,\ x \neq 0$, and $S = \{x \in R : f(x) = f(-x)\}$; then S:

(1) contains exactly one element.
(2) contains exactly two elements.
(3) contains more than two elements.
(4) is an empty set.

***Sol*.** **(2)**

$f(x) + 2f\left(\frac{1}{x}\right) = 3x$

replace x by $\frac{1}{x}$

$f\left(\frac{1}{x}\right) + 2f(x) = \frac{3}{x}$

$\Rightarrow f(x) = \frac{2}{x} - x$ as $f(x) = f(-x) \Rightarrow x = \pm\sqrt{2}$

36. Let $p = \lim_{x \to 0+} \left(1 + \tan^2\sqrt{x}\right)^{\frac{1}{2x}}$ then log p is equal to:

(1) 1
(2) $\frac{1}{2}$
(3) $\frac{1}{4}$
(4) 2.

***Sol*****.** **(2)**

$$p = \lim_{x\to 0^+}\left(1+\tan^2\sqrt{x}\right)^{\frac{1}{2x}}$$

$$= e^{\lim_{x\to 0^+}\frac{\tan^2\sqrt{x}}{2x}}$$

$$= e^{\lim_{x\to 0^+}\frac{1}{2}\left(\frac{\tan\sqrt{x}}{\sqrt{x}}\right)^2}$$

$$p = e^{\frac{1}{2}}$$

*37. A value of θ for which $\frac{2+3i\sin\theta}{1-2i\sin\theta}$ is purely imaginary, is:

(1) $\frac{\pi}{6}$ (2) $\sin^{-1}\left(\frac{\sqrt{3}}{4}\right)$

(3) $\sin^{-1}\left(\frac{1}{\sqrt{3}}\right)$ (4) $\frac{\pi}{3}$

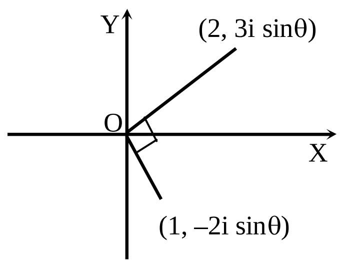
***Sol*****.** **(3)**

$\frac{2+3i\sin\theta}{1-2i\sin\theta}$ is purely imaginary $\in$ $\text{Arg}\left(\frac{2+3i\sin\theta}{1-2i\sin\theta}\right) = \frac{\pi}{2}, -\frac{\pi}{2}$

$\Rightarrow$ product of slopes taken as in xy plane is – 1

$\Rightarrow \frac{3\sin\theta}{2}\cdot\frac{-2\sin\theta}{1} = -1$

$\Rightarrow \sin^2\theta = \frac{1}{3},\ \theta = \sin^{-1}\left(\frac{1}{\sqrt{3}}\right).$




**Alternate Solution:**

$\frac{2+3i\sin\theta}{1-2i\sin\theta} = \frac{2-6\sin^2\theta+7i\sin\theta}{1+4\sin^2\theta}$ is purely imaginary

$\Rightarrow \frac{2-6\sin^2\theta}{1+4\sin^2\theta} = 0 \Rightarrow 6\sin^2\theta = 2.$

*38. The eccentricity of the hyperbola whose length of the latus rectum is equal to 8 and the length of its conjugate axis is equal half of the distance between its foci, is:

(1) $\frac{4}{\sqrt{3}}$ (2) $\frac{2}{\sqrt{3}}$

(3) $\sqrt{3}$ (4) $\frac{4}{3}$

***Sol*****.** **(2)**

Given, $\frac{2b^2}{a} = 8$ and $2b = \frac{1}{2}(2ae)$

$2b = ae$

$4b^2 = a^2 \cdot e^2$

$3e^2 = 4 \Rightarrow e = \frac{2}{\sqrt{3}}$

*39. If the standard deviation of the numbers 2, 3, a and 11 is 3.5, then which of the following is true?
(1) $3a^2 - 32a + 84 = 0$
(2) $3a^2 - 34a + 91 = 0$
(3) $3a^2 - 23a + 44 = 0$
(4) $3a^2 - 26a + 55 = 0$

***Sol.*** **(1)**

$$\bar{x} = \frac{2+3+a+11}{4} = \frac{a}{4} + 4$$

$$\sigma = \sqrt{\sum \frac{x_i^2}{n} - (\bar{x})^2}$$

$$3.5 = \sqrt{\frac{4+9+a^2+121}{4} - \left(\frac{a}{4}+4\right)^2}$$

$$\Rightarrow \frac{49}{4} = \frac{4\left(134+a^2\right) - \left(a^2+256+32a\right)}{16}$$

$\Rightarrow 3a^2 - 32a + 84 = 0$

40. The integral $\int \frac{2x^{12}+5x^9}{(x^5+x^3+1)^3} dx$ is equal to:
(1) $\frac{x^{10}}{2(x^5+x^3+1)^2} + C$
(2) $\frac{x^5}{2(x^5+x^3+1)^2} + C$
(3) $\frac{-x^{10}}{2(x^5+x^3+1)^2} + C$
(4) $\frac{-x^5}{(x^5+x^3+1)^2} + C$

where C is an arbitrary constant.

***Sol.*** **(1)**

$$I = \int \frac{\left(\frac{2}{x^3} + \frac{5}{x^6}\right)}{\left(1 + \frac{1}{x^2} + \frac{1}{x^5}\right)^3} dx, \quad \text{let } 1 + \frac{1}{x^2} + \frac{1}{x^5} = t$$

Hence $I = \frac{x^{10}}{2\left(x^5+x^3+1\right)^2} + c$.

41. If the line, $\frac{x-3}{2} = \frac{y+2}{-1} = \frac{z+4}{3}$ lies in the plane, $lx + my - z = 9$, then $l^2 + m^2$ is equal to:
(1) 18
(2) 5
(3) 2
(4) 26

***Sol.*** **(3)**

As line $\frac{x-3}{2} = \frac{y+2}{-1} = \frac{z+4}{3}$ lies in plane $\ell x + my - z = 9$

So $2\ell - m - 3 = 0$ (as line is perpendicular to normal of the plane) ..... (1)

Also point (3, –2, –4) lies in plane

So $3\ell - 2m - 5 = 0$ ..... (2)

From equation (1) and (2), we get $\ell = 1$, m = –1

So $\ell^2 + m^2 = 2$

*42. If $0 \le x < 2\pi$, then the number of real values of x, which satisfy the equation $\cos x + \cos 2x + \cos 3x + \cos 4x = 0$ is:

(1) 5 (2) 7
(3) 9 (4) 3

***Sol*.** **(2)**

$$2\cos\frac{5x}{2}\cdot\cos\frac{3x}{2}+2\cos\frac{5x}{2}\cdot\cos\frac{x}{2}=0$$

$$\Rightarrow \cos\frac{5x}{2}\cdot\left(2\cdot\cos x\cdot\cos\frac{x}{2}\right)=0$$

$$\Rightarrow \frac{x}{2}=(2n+1)\frac{\pi}{2},\quad x=(2m+1)\frac{\pi}{2},\quad \frac{5x}{2}=(2k+1)\frac{\pi}{2}\ ,\ (\text{where } n, m, k \in Z)$$

$$\Rightarrow x=(2n+1)\pi,\quad x=(2m+1)\frac{\pi}{2},\quad x=(2k+1)\frac{\pi}{5}$$

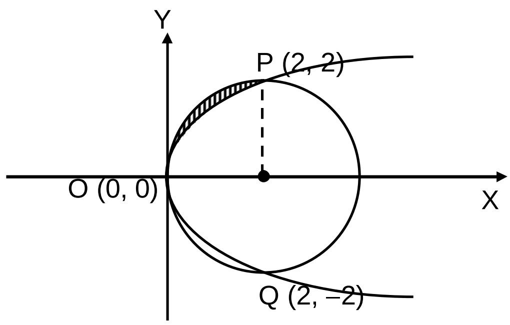
$$\Rightarrow x=\pi, \frac{\pi}{2}, \frac{3\pi}{2}, \frac{\pi}{5}, \frac{3\pi}{5}, \frac{7\pi}{5}, \frac{9\pi}{5}.$$

43. The area (in sq. units) of the region $\{(x, y): y^2 \ge 2x \text{ and } x^2+y^2 \le 4x,\ x \ge 0,\ y \ge 0\}$ is:

(1) $\pi-\frac{8}{3}$ (2) $\pi-\frac{4\sqrt{2}}{3}$
(3) $\frac{\pi}{2}-\frac{2\sqrt{2}}{3}$ (4) $\pi-\frac{4}{3}$

***Sol*.** **(1)**

The point of intersection of the curve $x^2 + y^2 = 4x$
$y^2 = 2x$ are (0, 0) and (2, 2) for $x \ge 0$ and $y \ge 0$




So required area = $\frac{1}{4}\pi\times 4-\int_0^2\sqrt{2x}\,dx$

$= \pi-\sqrt{2}\cdot\frac{2}{3}\left[x^{3/2}\right]_0^2$

$= \pi-\frac{8}{3}$

44. Let $\vec{a}, \vec{b}$ and $\vec{c}$ be three unit vectors such that $\vec{a}\times(\vec{b}\times\vec{c})=\frac{\sqrt{3}}{2}(\vec{b}+\vec{c})$. If $\vec{b}$ is not parallel to $\vec{c}$, then the angle between $\vec{a}$ and $\vec{b}$ is:

(1) $\frac{\pi}{2}$ (2) $\frac{2\pi}{3}$
(3) $\frac{5\pi}{6}$ (4) $\frac{3\pi}{4}$

***Sol*.** **(3)**

$$\vec{a}\times(\vec{b}\times\vec{c})=\frac{\sqrt{3}}{2}(\vec{b}+\vec{c})$$

$$\Rightarrow (\vec{a}\cdot\vec{c})\vec{b}-(\vec{a}\cdot\vec{b})\vec{c}=\frac{\sqrt{3}}{2}(\vec{b}+\vec{c})$$

$\Rightarrow\ \vec{a}\cdot\vec{c} = \frac{\sqrt{3}}{2}$ and $\vec{a}\cdot\vec{b} = -\frac{\sqrt{3}}{2}$

$\Rightarrow \cos\theta = -\frac{\sqrt{3}}{2}$ where θ is angle between $\vec{a}$ & $\vec{b}$

$\therefore\ \theta = \frac{5\pi}{6}$.

45. A wire of length 2 units is cut into two parts which are bent respectively to form a square of side = x units and a circle of radius = r units. If the sum of the areas of the square and the circle so formed is minimum, then:

(1) $(4-\pi)x = \pi r$  (2) $x = 2r$
(3) $2x = r$  (4) $2x = (\pi + 4)r$

***Sol.*** **(2)**

$f(x) = x^2 + \frac{(1-2x)^2}{\pi}$ (As $r = \frac{1-2x}{\pi}$)

$f'(x) = 2x - \frac{4(1-2x)}{\pi}$

$f''(x) = 2 + \frac{8}{\pi} > 0$

For minimum value of sum of area $f'(x) = 0$

$x = \frac{2}{\pi+4} \Rightarrow r = \frac{1}{\pi+4}$

$\Rightarrow x = 2r$.

46. The distance of the point (1, –5, 9) from the plane x – y + z = 5 measured along the line x = y = z is:

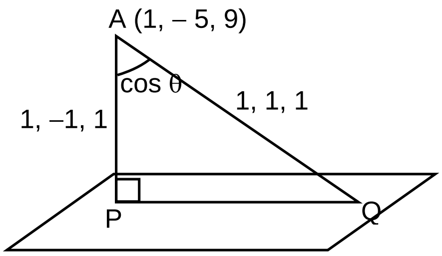
(1) $10\sqrt{3}$  (2) $\frac{10}{\sqrt{3}}$
(3) $\frac{20}{3}$  (4) $3\sqrt{10}$

***Sol.*** **(1)**

$\cos\theta = \frac{1-1+1}{3} = \frac{1}{3}$

$\cos\theta = \frac{AP}{AQ}$

$AQ = \frac{AP}{\cos\theta} = 10\sqrt{3}$




47. If a curve y = f(x) passes through the point (1, –1) and satisfies the differential equation, y(1 + xy) dx = x dy, then $f\left(-\frac{1}{2}\right)$ is equal to:

(1) $-\frac{4}{5}$  (2) $\frac{2}{5}$
(3) $\frac{4}{5}$  (4) $-\frac{2}{5}$

***Sol.*** **(3)**

y (1 + xy) dx = xdy

$$\frac{xdy - ydx}{y^2} = xdx$$

$$\int -d\left(\frac{x}{y}\right) = \int xdx$$

$$-\frac{x}{y} = \frac{x^2}{2} + c \text{ as } y(1) = -1 \Rightarrow c = \frac{1}{2}$$

Hence $y = \frac{-2x}{x^2+1} \Rightarrow f\left(-\frac{1}{2}\right) = \frac{4}{5}$.

*48. If the number of terms in the expansion of $\left(1-\frac{2}{x}+\frac{4}{x^2}\right)^n$, $x \neq 0$, is 28, then the sum of the coefficients of all the terms in this expansion, is:

(1) 2187 (2) 243

(3) 729 (4) 64

***Sol.*** **(3)**

Total number of terms $= {}^{n+2}C_2 = 28$

(n + 2)(n + 1) = 56

n = 6

Sum of coefficients = $(1 - 2 + 4)^n$

$= 3^6 = 729$

**[*Note: In the solution it is considered that different terms in the expansion having same powers are not merged, as such it should be a bonus question.]**

49. Consider

$f(x) = \tan^{-1}\left(\sqrt{\frac{1+\sin x}{1-\sin x}}\right), x \in \left(0, \frac{\pi}{2}\right)$. A normal to y = f(x) at $x = \frac{\pi}{6}$ also passes through the point:

(1) $\left(0, \frac{2\pi}{3}\right)$ (2) $\left(\frac{\pi}{6}, 0\right)$

(3) $\left(\frac{\pi}{4}, 0\right)$ (4) (0, 0)

***Sol.*** **(1)**

$f(x) = \tan^{-1}\sqrt{\frac{1+\sin x}{1-\sin x}}$, where $x \in \left(0, \frac{\pi}{2}\right)$

$= \tan^{-1}\left(\left|\tan\left(\frac{\pi}{4}+\frac{x}{2}\right)\right|\right)$

$\Rightarrow f(x) = \frac{\pi}{4} + \frac{x}{2}$, $\quad f\left(\frac{\pi}{6}\right) = \frac{\pi}{3}$

$f'(x) = \frac{1}{2}$

Equation of normal is

$$y - \frac{\pi}{3} = -2\left(x - \frac{\pi}{6}\right)$$

It passes through $\left(0, \frac{2\pi}{3}\right)$.

50. For $x \in \mathbf{R}$, $f(x) = |\log 2 - \sin x|$ and $g(x) = f(f(x))$, then:
(1) $g'(0) = \cos(\log 2)$
(2) $g'(0) = -\cos(\log 2)$
(3) g is differentiable at x = 0 and $g'(0) = -\sin(\log 2)$
(4) g is not differentiable at x = 0

***Sol.*** **(1)**
At x = 0, f is differential and $f'(0) = -\cos 0 = -1$
$g'(0) = f'(f(0))\cdot f'(0)$
$= -\cos(\log 2) \times -1$ (at x = 0, f(0) = log 2)
$= \cos(\log 2)$

51. Let two fair six–faced dice A and B be thrown simultaneously. If $E_1$ is the event that die A shows up four, $E_2$ is the event that die B shows up two and $E_3$ is the event that the sum of numbers on both dice is odd, then which of the following statements is **NOT true**?
(1) $E_2$ and $E_3$ are independent.
(2) $E_1$ and $E_3$ are independent.
(3) $E_1$, $E_2$ and $E_3$ are independent.
(4) $E_1$ and $E_2$ are independent.

***Sol.*** **(3)**
$P(E_1) = \frac{1}{6},\ P(E_2) = \frac{1}{6},\ P(E_3) = \frac{1}{2}$
Also $P(E_1 \cap E_2) = \frac{1}{36},\ P(E_2 \cap E_3) = \frac{1}{12},\ P(E_1 \cap E_3) = \frac{1}{12}$
And $P(E_1 \cap E_2 \cap E_3) = 0 \ne P(E_1)\cdot P(E_2)\cdot P(E_3)$
Hence, $E_1$, $E_2$, $E_3$ are not independent.

52. If $A = \begin{bmatrix} 5a & -b \\ 3 & 2 \end{bmatrix}$ and A adj A = A $A^T$, then 5a + b is equal to:
(1) 5
(2) 4
(3) 13
(4) – 1

***Sol.*** **(1)**
A adj A = $|A|\, I_n = \begin{bmatrix} 5a & -b \\ 3 & 2 \end{bmatrix}\begin{bmatrix} 5a & 3 \\ -b & 2 \end{bmatrix}$
$\Rightarrow (10a + 3b)\begin{bmatrix} 1 & 0 \\ 0 & 1 \end{bmatrix} = \begin{bmatrix} 25a^2 + b^2 & 15a - 2b \\ 15a - 2b & 13 \end{bmatrix}$
$\Rightarrow 15a - 2b = 0$ and $10a + 3b = 13$
$\Rightarrow 5a + b = 5 \times \frac{2}{5} + 3 = 5$.

*53. The Boolean Expression $(p \wedge {\sim}q) \vee q \vee ({\sim}p \wedge q)$ is equivalent to:
(1) $p \wedge q$
(2) $p \vee q$
(3) $p \vee {\sim} q$
(4) ${\sim} p \wedge q$

***Sol.*** **(2)**
$(p \wedge {\sim}q) \vee q \vee ({\sim}p \wedge q)$
$\equiv \{(p \vee q) \wedge ({\sim}q \vee q)\} \vee ({\sim}p \wedge q)$
$\equiv \{(p \vee q) \wedge T\} \vee ({\sim}p \wedge q)$
$\equiv (p \vee q) \vee ({\sim}p \wedge q)$
$\equiv \{(p \vee q) \vee {\sim}p\} \wedge (p \vee q \vee q)$
$\equiv T \wedge (p \vee q)$
$\equiv p \vee q$

*54. The sum of all real values of x satisfying the equation

$\left(x^2-5x+5\right)^{x^2+4x-60}=1$ is

(1) – 4 (2) 6
(3) 5 (4) 3

***Sol*.** **(4)**

Either $x^2 - 5x + 5 = 1$ or $x^2 + 4x - 60 = 0$
x = 1, 4 or x = –10, 6
Also $x^2 - 5x + 5 = -1$ and $x^2 + 4x - 60 \in$ even number
x = 2, 3
For x = 3 $x^2 + 4x - 60$ is odd
Total solutions are x = 1, 4, –10, 6, 2
$\Rightarrow$ Sum = 3

*55. The centres of those circles which touch the circle, $x^2 + y^2 - 8x - 8y - 4 = 0$, externally and also touch the x–axis, lie on:

(1) an ellipse which is not a circle. (2) a hyperbola.
(3) a parabola. (4) a circle.

***Sol*.** **(3)**

Let (h, k) be the centre of the circle which touch x–axis and $x^2 + y^2 - 8x - 8y - 4 = 0$ externally.
$\Rightarrow$ Radius of that circle is |k|
$\Rightarrow$ $(h-4)^2 + (k-4)^2 = (|k| + 6)^2$
$\Rightarrow$ $x^2 - 8x - 20y - 4 = 0$ if $y \geq 0$
and $x^2 - 8x + 4y - 4 = 0$ if $y < 0$
$\Rightarrow$ The curve is parabola.

*56. If all the words (with or without meaning) having five letters, formed using the letters of the word SMALL and arranged as in a dictionary; then the position of the word SMALL is:

(1) $59^{th}$ (2) $52^{nd}$
(3) $58^{th}$ (4) $46^{th}$

***Sol*.** **(3)**

Words starting with A, L, M = $\frac{4!}{2!}+4!+\frac{4!}{2!}$ = 48

Words starting with SA, SL = $\frac{3!}{2!}$ + 3! = 9

$\Rightarrow$ Rank of the word SMALL = 58.

57. $\lim\limits_{n\to\infty}\left(\frac{(n+1)(n+2)...3n}{n^{2n}}\right)^{1/n}$ is equal to:

(1) $\frac{27}{e^2}$ (2) $\frac{9}{e^2}$

(3) 3 log3 – 2 (4) $\frac{18}{e^4}$

***Sol*.** **(1)**

$$\ln y = \lim_{n\to\infty}\frac{1}{n}\cdot\log\left(1+\frac{1}{n}\right)\left(1+\frac{2}{n}\right)...\left(1+\frac{2n}{n}\right)$$

$$=\frac{1}{n}\cdot\sum_{r=1}^{2n}\ln\left(1+\frac{r}{n}\right)$$

$\ln y = \int_0^2 \ln(1+x)\,dx$, let t = 1 + x

$= \int_1^3 \ln t \, dt$

$= \ln \frac{27}{e^2}$

*58. If the sum of the first ten terms of the series $\left(1\frac{3}{5}\right)^2 + \left(2\frac{2}{5}\right)^2 + \left(3\frac{1}{5}\right)^2 + 4^2 + \left(4\frac{4}{5}\right)^2 + ....$, is $\frac{16}{5}m$, then m is equal to:

(1) 101 (2) 100
(3) 99 (4) 102

***Sol.*** **(1)**

Let $S = \left(\frac{8}{5}\right)^2 + \left(\frac{12}{5}\right)^2 + \left(\frac{16}{5}\right)^2 + \left(\frac{20}{5}\right)^2 + ..... + \left(\frac{44}{5}\right)^2$

$\Rightarrow S = \frac{16}{25}\left[2^2 + 3^2 + 4^2 + 5^2 ..... + 11^2\right]$

$\Rightarrow S = \frac{16}{25}\left[1^2 + 2^2 ... + 11^2 - 1\right] = \frac{16}{5} \times 101$

$\Rightarrow m = 101$

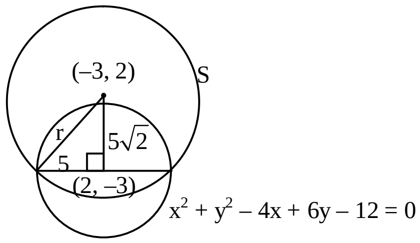
*59. If one of the diameters of the circle, given by the equation, $x^2 + y^2 - 4x + 6y - 12 = 0$, is a chord of a circle S, whose centre is at (–3, 2), then the radius of S is:

(1) $5\sqrt{3}$ (2) 5
(3) 10 (4) $5\sqrt{2}$

***Sol.*** **(1)**

Let 'r' be the radius of circle S

$\Rightarrow r = 5\sqrt{3}$




*60. A man is walking towards a vertical pillar in a straight path, at a uniform speed. At a certain point A on the path, he observes that the angle of elevation of the top of the pillar is 30º. After walking for 10 minutes from A in the same direction, at a point B, he observes that the angle of elevation of the top of the pillar is 60º. Then the time taken (in minutes) by him, from B to reach the pillar, is:

(1) 10 (2) 20
(3) 5 (4) 6

***Sol.*** **(3)**

$\tan 60° = \frac{h}{x} \Rightarrow h = \sqrt{3}x$

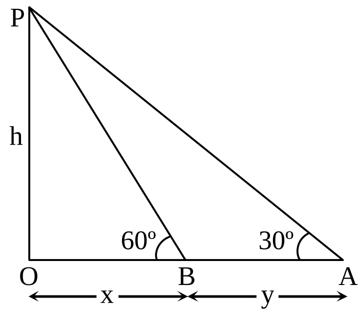
$\tan 30° = \frac{h}{x + y} \Rightarrow \sqrt{3}h = x + y$

$3x = x + y$

$\Rightarrow 2x = y$

Time taken from A to B is 10 min

So time taken from B to pillar is 5 min



# PART C – PHYSICS

*61. A uniform string of length 20 m is suspended from a rigid support. A short wave pulse is introduced at its lowest end. It starts moving up the string. The time taken to reach the support is : (take g = 10 $ms^{-2}$)

(1) 2s (2) $2\sqrt{2}$ s

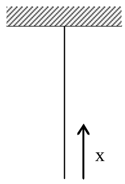
(3) $\sqrt{2}$ s (4) $2\pi\sqrt{2}$ s

***Sol.*** **(2)**

$T(x) = \frac{Mgx}{L}$

$\Rightarrow v(x) = \sqrt{\frac{T}{\mu}} = \sqrt{gx}$

$\Rightarrow \frac{dx}{dt} = \sqrt{gx} \Rightarrow$ time taken $= 2\sqrt{2}$ s .




*62. A person trying to lose weight by burning fat lifts a mass of 10 kg upto a height of 1 m 1000 times. Assume that the potential energy lost each time he lowers the mass is dissipated. How much fat will he use up considering the work done only when the weight is lifted up? Fat supplies $3.8 \times 10^7$ J of energy per kg which is converted to mechanical energy with a 20 % efficiency rate. Take g = 9.8 $ms^{-2}$ :

(1) $6.45 \times 10^{-3}$ kg (2) $9.89 \times 10^{-3}$ kg

(3) $12.89 \times 10^{-3}$ kg (4) $2.45 \times 10^{-3}$ kg

***Sol.*** **(3)**

Let fat used be 'x' kg

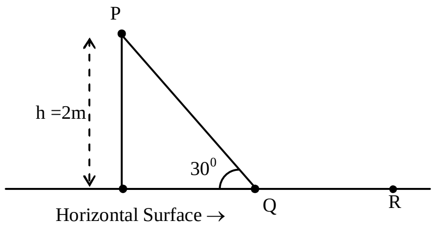
$\Rightarrow$ Mechanical energy available = $x \times 3.8 \times 10^7 \times \frac{20}{100}$

Work done in lifting up = $10 \times 9.8 \times 1000$

$\Rightarrow x \times 3.8 \times 10^7 \times \frac{20}{100} = 9.8 \times 10^4$

$\Rightarrow x \approx 12.89 \times 10^{-3}$ kg.

*63. A point particle of mass m, moves along the uniformly rough track PQR as shown in the figure. The coefficient of friction, between the particle and the rough track equals $\mu$. The particle is released, from rest, from the point P and it comes to rest at a point R. The energies, lost by the ball, over the parts, PQ and QR, of the track, are equal to each other, and no energy is lost when particle changes direction from PQ to QR.




The values of the coefficient of friction $\mu$ and the distance x(=QR), are, respectively close to :

(1) 0.2 and 3.5 m (2) 0.29 and 3.5 m

(3) 0.29 and 6.5 m (4) 0.2 and 6.5 m

***Sol.*** **(2)**

Since work done by friction on parts PQ and QR are equal

$-\mu mg \times \frac{\sqrt{3}}{2} \times 4 = -\mu mgx$ (QR = x)

$\Rightarrow x = 2\sqrt{3}$ m $\approx 3.5$ m

Applying work energy theorem from P to R

$mg \sin 30^0 \times 4 - \mu mg \frac{\sqrt{3}}{2} \times 4 - \mu mgx = 0$

$\Rightarrow \mu = \frac{1}{2\sqrt{3}} \approx 0.29.$

64. Two identical wires A and B, each of length '$\ell$', carry the same current I. Wire A is bent into a circle of radius R and wire B is bent to form a square of side 'a'. If $B_A$ and $B_B$ are the values of magnetic field at the centres of the circle and square respectively, then the ratio $\frac{B_A}{B_B}$ is :

(1) $\frac{\pi^2}{16\sqrt{2}}$ (2) $\frac{\pi^2}{16}$

(3) $\frac{\pi^2}{8\sqrt{2}}$ (4) $\frac{\pi^2}{8}$

***Sol.*** **(3)**

$B_A = \frac{\mu_0}{4\pi}\frac{2\pi i}{(\ell/2\pi)}$

$B_B = \left[\frac{\mu_0}{4\pi}\frac{i}{\ell/8}(\sin 45^0 + \sin 45^0)\right] \times 4$

$\frac{B_A}{B_B} = \frac{\pi^2}{8\sqrt{2}}$

65. A galvanometer having a coil resistance of 100 Ω gives a full scale deflection, when a current of 1 mA is passed through it. The value of the resistance, which can convert this galvanometer into ammeter giving a full scale deflection for a current of 10 A is :

(1) 2Ω (2) 0.1 Ω

(3) 3 Ω (4) 0.01 Ω

***Sol.*** **(4)**

For full scale deflection

$100 \times i_g = (i - i_g)S$

where 'S' is the required resistance

$S = \frac{100 \times 1 \times 10^{-3}}{(10 - 10^{-3})}$

$S \approx 0.01\ \Omega$

66. An observer looks at a distant tree of height 10 m with a telescope of magnifying power of 20. To the observer the tree appears :

(1) 10 times nearer. (2) 20 times taller.

(3) 20 times nearer. (4) 10 times taller.

***Sol.*** **(3)**

67. The temperature dependence of resistances of **Cu** and undoped **Si** in the temperature range 300 – 400 K, is best described by :

(1) Linear increase for Cu, exponential increase for Si.

(2) Linear increase for Cu, exponential decrease for Si.

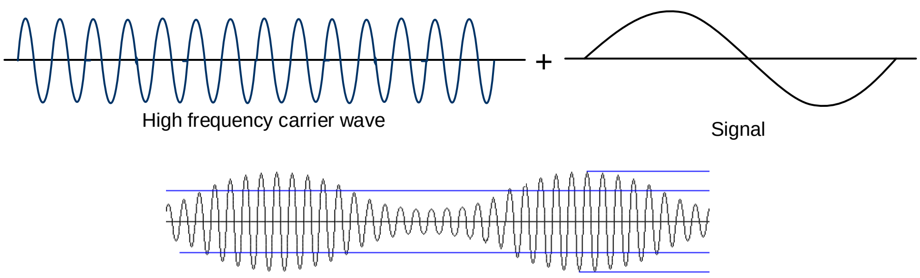
(3) Linear decrease for Cu, linear decrease for Si.

(4) Linear increase for Cu, linear increase for Si.

***Sol.*** **(2)**
The electric resistance of a typical intrinsic (undoped) semiconductor decreases exponentially with temperature
$\rho = \rho_0 e^{-aT}$
where a is a constant.

68. Choose the correct statement :
(1) In amplitude modulation the frequency of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal.
(2) In frequency modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal.
(3) In frequency modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the frequency of the audio signal.
(4) In amplitude modulation the amplitude of the high frequency carrier wave is made to vary in proportion to the amplitude of the audio signal.

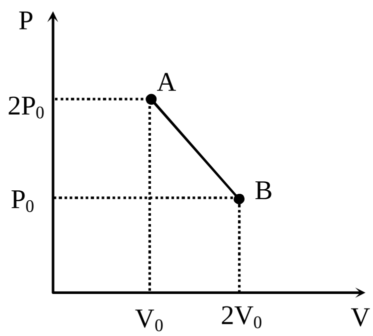
***Sol.*** **(4)**
In amplitude modulation amplitude of carrier wave varies in proportion to applied signal.




69. Half-lives of two radioactive elements A and B are 20 minutes and 40 minutes, respectively. Initially, the samples have equal number of nuclei. After 80 minutes, the ratio of decayed numbers of A and B nuclei will be :
(1) 4 : 1 (2) 1 : 4
(3) 5 : 4 (4) 1 : 16

***Sol.*** **(3)**
for A : $N_0 - \frac{N_0}{2^4} = \frac{15N_0}{16}$
for B : $N_0 - \frac{N_0}{2^2} = \frac{3N_0}{4}$
The required ratio is $\frac{5}{4}$

*70. 'n' moles of an ideal gas undergoes a process A → B as shown in the figure. The maximum temperature of the gas during the process will be :




(1) $\frac{3P_0V_0}{2nR}$
(2) $\frac{9P_0V_0}{2nR}$
(3) $\frac{9P_0V_0}{nR}$
(4) $\frac{9P_0V_0}{4nR}$

***Sol.*** **(4)**

Equation of line is

$PV_0 + P_0V = 3P_0V_0$ …(i)

Also PV = nRT …(ii)

for $T_{max}, \frac{dT}{dV} = 0$

$\Rightarrow V = \frac{3V_0}{2}, \ P = \frac{3P_0}{2}$

$\Rightarrow T_{max} = \frac{9P_0V_0}{4nR}$

71. An arc lamp requires a direct current of 10 A and 80 V to function. If it is connected to a 220 V (rms), 50 Hz AC supply, the series inductor needed for it to work is close to :

(1) 0.08 H (2) 0.044 H

(3) 0.065 H (4) 80 H

***Sol.*** **(3)**

For the lamp with direct current,

V = IR

$\Rightarrow R = 8\Omega$ and $P = 80 \times 10 = 800$ W

For ac supply

$P = I_{rms}^2 R = \frac{\varepsilon_{rms}^2}{Z^2} R$

$\Rightarrow Z^2 = \frac{(220)^2 \times 8}{800}$

$\Rightarrow Z = 22\Omega$

$\Rightarrow R^2 + \omega^2 L^2 = (22)^2$

$\Rightarrow \omega L = \sqrt{420}$

$\Rightarrow L = 0.065$ H

*72. A pipe open at both ends has a fundamental frequency *f* in air. The pipe is dipped vertically in water so that half of it is in water. The fundamental frequency of the air column is now :

(1) $\frac{3f}{4}$ (2) 2f

(3) f (4) $\frac{f}{2}$

***Sol.*** **(3)**

For open pipe in air, fundamental frequency:

$f = \frac{V}{2\ell}$

For the pipe closed at one end (dipped in water), fundamental frequency:

$f' = \frac{V}{4\ell'} = \frac{V}{4 \times \frac{\ell}{2}} = \frac{V}{2\ell}$

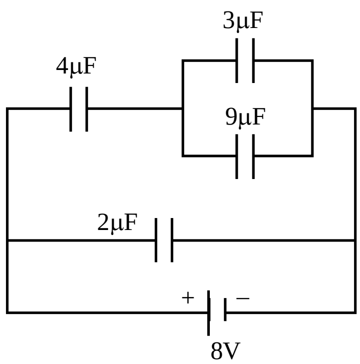
$\therefore \quad f' = f$

73. The box of pin hole camera, of length L, has a hole of radius *a*. It is assumed that when the hole is illuminated by a parallel beam of light of wavelength $\lambda$ the spread of the spot (obtained on the opposite wall of the camera) is the sum of its geometrical spread and the spread due to diffraction. The spot would then have its minimum size (say $b_{min}$) when :

(1) $a = \sqrt{\lambda L}$ and $b_{min} = \left(\frac{2\lambda^2}{L}\right)$
(2) $a = \sqrt{\lambda L}$ and $b_{min} = \sqrt{4\lambda L}$
(3) $a = \frac{\lambda^2}{L}$ and $b_{min} = \sqrt{4\lambda L}$
(4) $a = \frac{\lambda^2}{L}$ and $b_{min} = \left(\frac{2\lambda^2}{L}\right)$

***Sol.*** **(2)**
We know that
Geometrical spread = a
and diffraction spread = $\frac{\lambda L}{a}$
so spot size(b) = $a + \frac{\lambda L}{a}$
for minimum spot size $a = \frac{\lambda L}{a}$
$\Rightarrow \quad a = \sqrt{\lambda L}$
and $b_{min} = \sqrt{\lambda L} + \sqrt{\lambda L} = \sqrt{4\lambda L}$

74. A combination of capacitors is set up as shown in the figure. The magnitude of the electric field, due to a point charge Q (having a charge equal to the sum of the charges on the 4 μF and 9 μF capacitors), at a point distant 30 m from it, would equal :



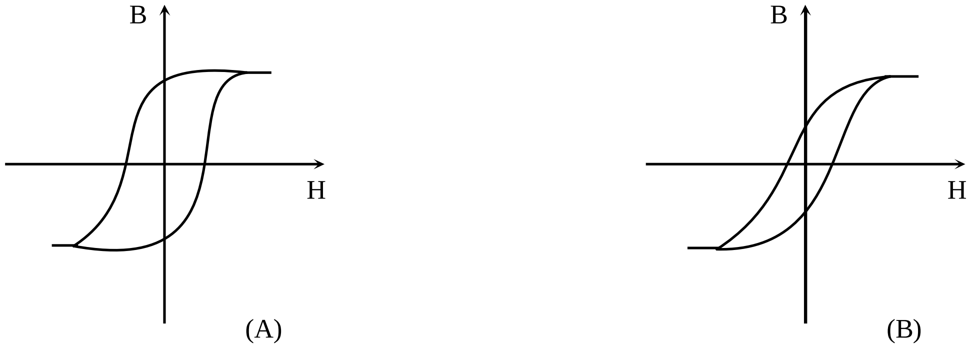

(1) 360 N/C
(2) 420 N/C
(3) 480 N/C
(4) 240 N/C

***Sol.*** **(2)**
Charge on 9 μF capacitor = 18 μC
Charge on 4 μF capacitor = 24 μC
$\therefore$ Q = 24 + 18 = 42 μC
$\therefore \quad \frac{KQ}{r^2} = \frac{9\times10^9 \times 42\times10^{-6}}{(30)^2} = 420$ N/C

75. Arrange the following electromagnetic radiations per quantum in the order of increasing energy :
A : Blue light B : Yellow light
C : X-ray D : Radiowave
(1) A, B, D, C
(2) C, A, B, D
(3) B, A, D, C
(4) D, B, A, C

***Sol.*** **(4)**
Radiation energy per quantum is
E = hν
As per EM spectrum, the increasing order of frequency and hence energy is
Radio wave < Yellow light < Blue light < X Ray

76. Hysteresis loops for two magnetic materials A and B are given below :



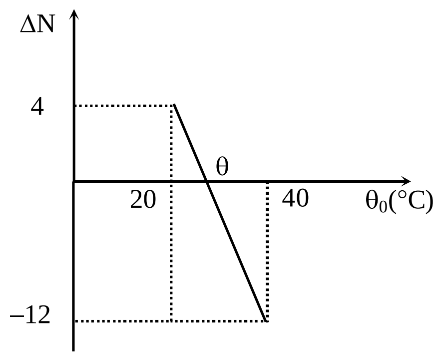
(A) (B)

These materials are used to make magnets for electric generators, transformer core and electromagnet core. Then it is proper to use :
(1) A for electromagnets and B for electric generators
(2) A for transformers and B for electric generators
(3) B for electromagnets and transformers
(4) A for electric generators and transformers

***Sol.*** **(3)**
For electromagnet and transformer, the coercivity should be low to reduce energy loss

*77. A pendulum clock loses 12 s a day if the temperature is 40°C and gains 4 s a day if the temperature is 20°C. The temperature at which the clock will show correct time, and the co-efficient of linear expansion ($\alpha$) of the metal of the pendulum shaft are respectively :
(1) 60°C; $\alpha = 1.85 \times 10^{-4}$/°C
(2) 30°C; $\alpha = 1.85 \times 10^{-3}$/°C
(3) 55°C; $\alpha = 1.85 \times 10^{-2}$/°C
(4) 25°C; $\alpha = 1.85 \times 10^{-5}$/°C

***Sol.*** **(4)**
**First Method**




Time period T = $2\pi\sqrt{\frac{\ell}{g}} = 2\pi\sqrt{\frac{\ell_0}{g}(1+\alpha\Delta\theta)}$

$$T = T_0\left[1+\frac{1}{2}\alpha\Delta\theta\right]$$

$$N = \frac{2\times 86400}{T} = \left(\frac{2\times 86400}{T_0}\right)\left(1+\frac{1}{2}\alpha\Delta\theta\right) = N\left(1+\frac{\alpha\Delta\theta}{2}\right)$$

$$\Delta N = N - N_0 = \frac{1}{2}\alpha\Delta\theta N_0 \Rightarrow \Delta N \propto \Delta\theta$$

$\Rightarrow \theta_0 = 25$°C
Putting $\theta_0$ , we get $\alpha = 1.85 \times 10^{-5}$/°C
**Second Method**
According to given conditions

$$86412 = 2\pi\sqrt{\frac{\ell_{40}}{g}} \quad \text{...(i)}$$

$$86396 = 2\pi\sqrt{\frac{\ell_{20}}{g}} \quad \text{...(ii)}$$

$$86400 = 2\pi\sqrt{\frac{\ell}{g}} \quad \text{...(iii)}$$

From equation (i) and (iii)

$12 = \frac{2\pi}{\sqrt{g}}\left[\sqrt{\ell_{40}} - \sqrt{\ell}\right]$ …(iv)

and $4 = \frac{2\pi}{\sqrt{g}}\left[\sqrt{\ell} - \sqrt{\ell_{20}}\right]$ …(v)

on dividing (iv) and (v)

$$3 = \frac{\sqrt{1+\alpha(40-\theta)}-1}{1-\sqrt{1+\alpha(20-\theta)}}$$

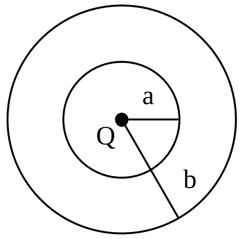
$\Rightarrow 3 = \frac{40-\theta}{\theta-20}$ (by Binomial theorem)

$\Rightarrow \theta = 25°$

on using θ in (i) and (iii)

$\frac{86412}{86400} = \sqrt{1+\alpha 15} \Rightarrow \alpha = 1.85 \times 10^{-5}/°C$

78. The region between two concentric spheres of radii 'a' and 'b', respectively (see figure), has volume charge density $\rho = \frac{A}{r}$, where A is a constant and r is the distance from the centre. At the centre of the spheres is a point charge Q. The value of A such that the electric field in the region between the spheres will be constant, is :




(1) $\frac{Q}{2\pi(b^2-a^2)}$ (2) $\frac{2Q}{\pi(a^2-b^2)}$

(3) $\frac{2Q}{\pi a^2}$ (4) $\frac{Q}{2\pi a^2}$

***Sol.*** **(4)**

According Gauss's law, we can write,

$$E \times 4\pi r^2 = \frac{1}{\varepsilon_0}\left[Q + \int_a^r \left(\frac{A}{r}\right)\left(4\pi r^2 dr\right)\right] = \frac{1}{\varepsilon_0}\left[Q + 2\pi A(r^2 - a^2)\right]$$

For E to be independent of r,

$Q - 2\pi A a^2 = 0 \Rightarrow a = \frac{Q}{2\pi a^2}$

79. In an experiment for determination of refractive index of glass of a prism by $i - \delta$, plot, it was found that a ray incident at angle 35°, suffers a deviation of 40° and that it emerges at angle 79°. In that case which of the following is closest to the maximum possible value of the refractive index ?

(1) 1.6 (2) 1.7

(3) 1.8 (4) 1.5

***Sol.*** **(4)**

$\delta = i + e - A \Rightarrow A = 74°$

$$\mu = \frac{\sin\left(\frac{A+\delta_{min}}{2}\right)}{\sin\left(\frac{A}{2}\right)} = \frac{5}{3}\sin\left(37° + \frac{\delta_{min}}{2}\right)$$

$\mu_{max}$ can be $\frac{5}{3}$, so $\mu$ will be less than $\frac{5}{3}$

Since $\delta_{min}$ will be less than 40°, so

$\mu < \frac{5}{3}\sin 57° < \frac{5}{3}\sin 60° \Rightarrow \mu < 1.446$

So the nearest possible value of $\mu$ should be 1.5

80. A student measures the time period of 100 oscillations of a simple pendulum four times. The data set is 90s, 91s, 95s and 92s. If the minimum division in the measuring clock is 1s, then the reported mean time should be :

(1) $92 \pm 5.0$s (2) $92 \pm 1.8$s
(3) $92 \pm 3$s (4) $92 \pm 2$s

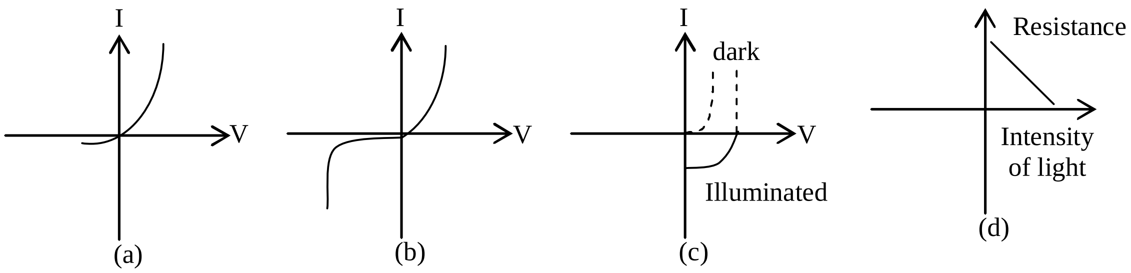
***Sol.*** **(4)**

$t = \frac{t_1 + t_2 + t_3 + t_4}{4} = \frac{90+91+95+92}{4} = 92$

Now mean deviation is equal to $\left(\frac{2+1+3+0}{4}\right) = 1.5$

Since least count of clock is one second, so $\Delta t = 2\sec$

81. Identify the semiconductor devices whose characteristics are given below, in the order (a), (b), (c), (d) :




(1) Zener diode, Simple diode, Light dependent resistance, Solar cell
(2) Solar cell, Light dependent resistance, Zener diode, Simple diode
(3) Zener diode, Solar cell, Simple diode, Light dependent resistance
(4) Simple diode, Zener diode, Solar cell, Light dependent resistance

***Sol.*** **(4)**

82. Radiation of wavelength λ, is incident on a photocell. The fastest emitted electron has speed v. If the wavelength is changed to $\frac{3\lambda}{4}$, the speed of the fastest emitted electron will be :

(1) $< v\left(\frac{4}{3}\right)^{\frac{1}{2}}$ (2) $= v\left(\frac{4}{3}\right)^{\frac{1}{2}}$

(3) $= v\left(\frac{3}{4}\right)^{\frac{1}{2}}$ (4) $> v\left(\frac{4}{3}\right)^{\frac{1}{2}}$

***Sol.*** **(4)**

$\frac{hc}{\lambda} - \phi = \frac{1}{2}mv^2$ ...(i)

$\frac{4hc}{3\lambda} - \phi = \frac{1}{2}mv_1^2$ ...(ii)

So, $\frac{hc}{3\lambda} = \frac{1}{2}m(v_1^2 - v^2)$

$$\frac{1}{3}\left(\frac{1}{2}mv^2+\phi\right)=\frac{1}{2}m(v_1^2-v^2)$$

$$\therefore \quad v_1 > v\left(\frac{4}{3}\right)^{\frac{1}{2}}$$

*83. A particle performs simple harmonic motion with amplitude A. Its speed is trebled at the instant that it is at a distance $\frac{2A}{3}$ from equilibrium position. The new amplitude of the motion is :

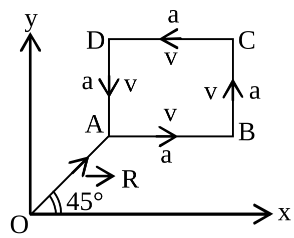
(1) 3A
(2) $A\sqrt{3}$
(3) $\frac{7A}{3}$
(4) $\frac{A}{3}\sqrt{41}$

***Sol.*** **(3)**

$$3\omega\sqrt{A^2-\left(\frac{2A}{3}\right)^2}=\omega\sqrt{A_1^2-\left(\frac{2A}{3}\right)^2}$$

$$\therefore \quad A_1=\frac{7A}{3}$$

*84. A particle of mass m is moving along the side of a square of side 'a', with a uniform speed v in the x-y plane as shown in the figure :
Which of the following statements is **false** for the angular momentum $\vec{L}$ about the origin ?




(1) $\vec{L}=mv\left[\frac{R}{\sqrt{2}}-a\right]\hat{k}$ when the particle is moving from C to D.

(2) $\vec{L}=mv\left[\frac{R}{\sqrt{2}}+a\right]\hat{k}$ when the particle is moving from B to C.

(3) $\vec{L}=\frac{mv}{\sqrt{2}}R\,\hat{k}$ when the particle is moving from D to A.

(4) $\vec{L}=-\frac{mv}{\sqrt{2}}R\,\hat{k}$ when the particle is moving from A to B.

***Sol.*** **(1, 3)**

$$\vec{L}_O=mv\frac{R}{\sqrt{2}}(-\hat{k}) \quad \text{[D to A]}$$

$$\vec{L}_O=mv\left[\frac{R}{\sqrt{2}}+a\right]\hat{k} \quad \text{[C to D]}$$

*85. An ideal gas undergoes a quasi static, reversible process in which its molar heat capacity C remains constant. If during this process the relation of pressure P and volume V is given by $PV^n$ = constant, then n is given by (Here $C_P$ and $C_V$ are molar specific heat at constant pressure and constant volume, respectively):

(1) $n=\frac{C-C_P}{C-C_V}$
(2) $n=\frac{C_P-C}{C-C_V}$
(3) $n=\frac{C-C_V}{C-C_P}$
(4) $n=\frac{C_P}{C_V}$

***Sol.*** **(1)**

$$C = C_V - \frac{R}{n-1}$$

$$\frac{R}{n-1} = C_V - C$$

$$n = \frac{C - C_P}{C - C_V}$$

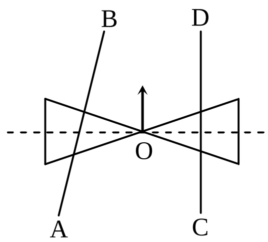
86. A screw gauge with a pitch of 0.5 mm and a circular scale with 50 divisions is used to measure the thickness of a thin sheet of Aluminium. Before starting the measurement, it is found that when the two jaws of the screw gauge are brought in contact, the $45^{th}$ division coincides with the main scale line and that the zero of the main scale is barely visible. What is the thickness of the sheet if the main scale reading is 0.5 mm and the $25^{th}$ division coincides with the main scale line ?

(1) 0.80 mm (2) 0.70 mm
(3) 0.50 mm (4) 0.75 mm

***Sol.*** **(1)**

Reading = $0.5 + 25\left(\frac{0.5}{50}\right) + 5\left(\frac{0.5}{50}\right) = 0.8\,\text{mm}$

*87. A roller is made by joining together two cones at their vertices O. It is kept on two rails AB and CD which are placed asymmetrically (see figure), with its axis perpendicular to CD and its centre O at the centre of line joining AB and CD (see figure). It is given a light push so that it starts rolling with its centre O moving parallel to CD in the direction shown. As it moves, the roller will tend to :

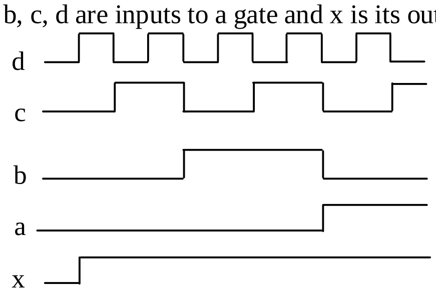



(1) turn right (2) go straight
(3) turn left and right alternately (4) turn left

***Sol.*** **(4)**

From normal reactions of roller, we can conclude it moves towards left.

88. If a, b, c, d are inputs to a gate and x is its output, then, as per the following time graph, the gate is :




(1) AND (2) OR
(3) NAND (4) NOT

***Sol.*** **(2)**

From truth Table, it is OR gate.

89. For a common emitter configuration, if $\alpha$ and $\beta$ have their usual meanings, the incorrect relationship between $\alpha$ and $\beta$ is :

(1) $\alpha = \frac{\beta}{1-\beta}$ (2) $\alpha = \frac{\beta}{1+\beta}$

(3) $\alpha = \frac{\beta^2}{1+\beta^2}$ (4) $\frac{1}{\alpha} = \frac{1}{\beta} + 1$

***Sol.*** **(1, 3)**

$\beta = \frac{\alpha}{1-\alpha}$, is not satisfied by option (1, 3)

*90. A satellite is revolving in a circular orbit at a height 'h' from the earth's surface (radius of earth R; h<<R). The minimum increase in its orbital velocity required, so that the satellite could escape from the earth's gravitational field, is close to : (Neglect the effect of atmosphere.)

(1) $\sqrt{gR}$ (2) $\sqrt{gR/2}$

(3) $\sqrt{gR}\left(\sqrt{2}-1\right)$ (4) $\sqrt{2gR}$

***Sol.*** **(3)**

At height h escape velocity

$V_e = \sqrt{\frac{2GM}{h+h}}$

Orbital velocity $V_0 = \sqrt{\frac{GM}{R+h}}$

$\therefore$ Increase in orbital velocity required to escape gravitational field

$\Rightarrow$ $V_e - V_0$

$\Rightarrow$ $\sqrt{gR}\left(\sqrt{2}-1\right)$

***Read the following instructions carefully:***

1. The candidates should fill in the required particulars on the Test Booklet and Answer Sheet (*Side–1*) with ***Blue/Black Ball Point Pen.***

2. For writing/marking particulars on ***Side–2*** of the Answer Sheet, use ***Blue/Black Ball Point Pen only.***

3. The candidates should not write their Roll Numbers anywhere else (except in the specified space) on the Test Booklet/Answer Sheet.

4. Out of the four options given for each question, only one option is the correct answer.

5. For each ***incorrect response, one-fourth (¼)*** of the total marks allotted to the **question** would be deducted from the total score. ***No deduction*** from the total score, however, will be made ***if no response*** is indicated for an item in the Answer Sheet.

6. Handle the Test Booklet and Answer Sheet with care, *as under no circumstances* (*except for discrepancy in Test Booklet Code and Answer Sheet Code), another set will be provided.*

7. The candidates are not allowed to do any rough work or writing work on the Answer sheet. All calculations/writing work are to be done in the space provided for this purpose in the Test Booklet itself, marked 'Space for Rough Work'. This space is given at the bottom of each page and in **three** pages (Pages 21 – 23) at the end of the booklet.

8. On completion of the test, the candidates must hand over the Answer Sheet to the Invigilator on duty in the Room/Hall. **However, the candidates are allowed to take away this Test Booklet with them.**

9. Each candidate must show on demand his/her Admit Card to the Invigilator.

10. No candidate, without special permission of the Superintendent or Invigilator, should leave his/her seat.

11. The candidates should not leave the Examination Hall without handing over their Answer Sheet to the Invigilator on duty and sign the Attendance Sheet again. Cases where a candidate has not signed the Attendance Sheet a second time will be deemed not to have handed over the Answer Sheet and dealt with as an unfair means case. **The candidates are also required to put their left hand THUMB impression in the space provided in the Attendance Sheet.**

12. Use of Electronic/Manual Calculator and any Electronic device like mobile phone, pager etc. is prohibited.

13. The candidates are governed by all Rules and Regulations of the JAB/Board with regard to their conduct in the Examination Hall. All cases of unfair means will be dealt with as per Rules and Regulations of the JAB/Board.

14. No part of the Test Booklet and Answer Sheet shall be detached under any circumstance.

15. **Candidates are not allowed to carry any textual material, printed or written, bits of papers, pager, mobile phone, electronic device or any other material except the Admit Card inside the examination room/hall.**